

# ***Potency of Greenhouse Molecules***

**32<sup>nd</sup> Conference Climate Variability/Change  
American Meteorological Society**



**Phoenix, USA**

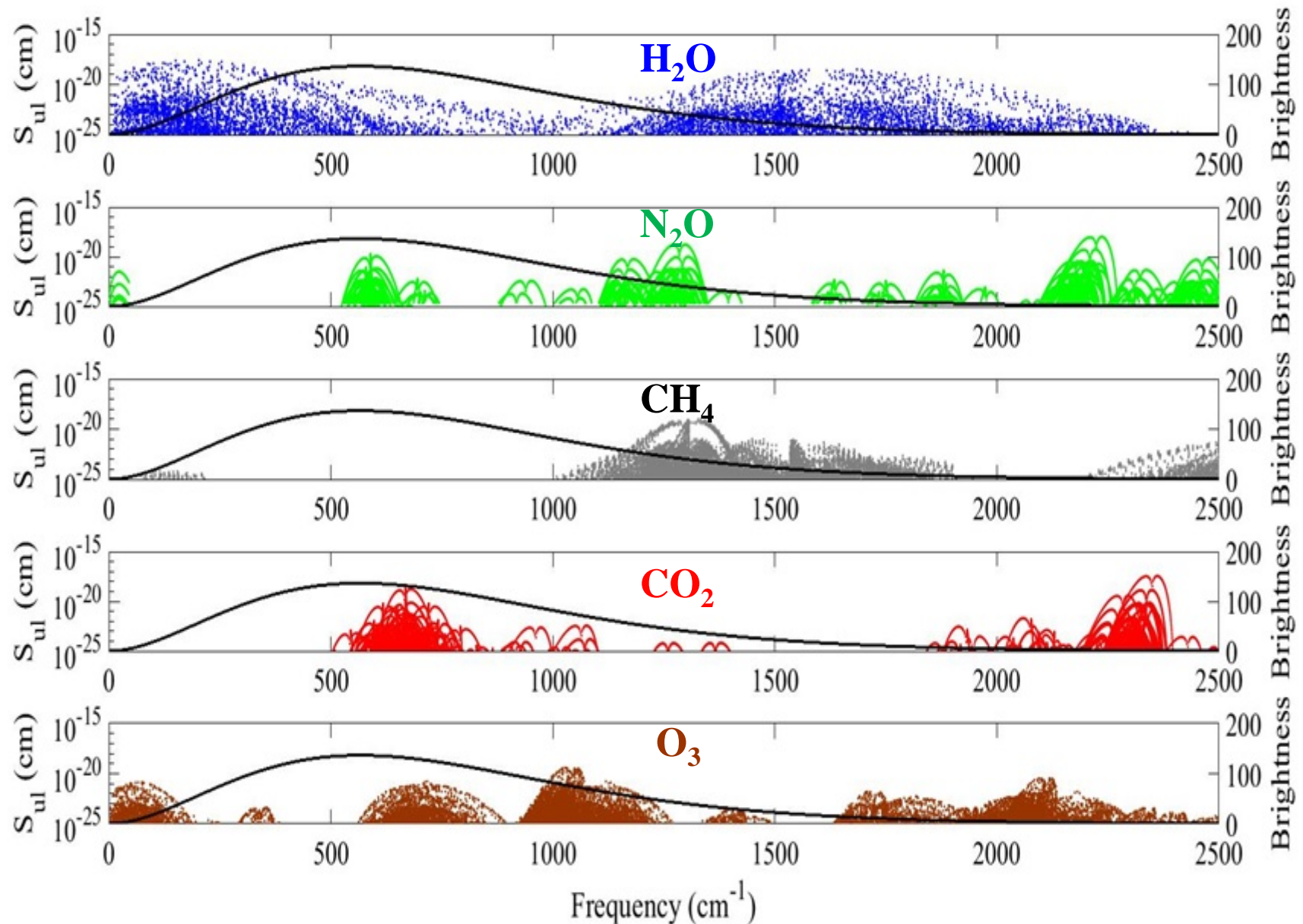
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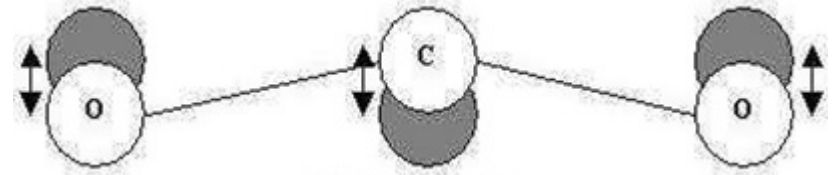
# Greenhouse Gas Transitions

Line Strengths from Hitran Database

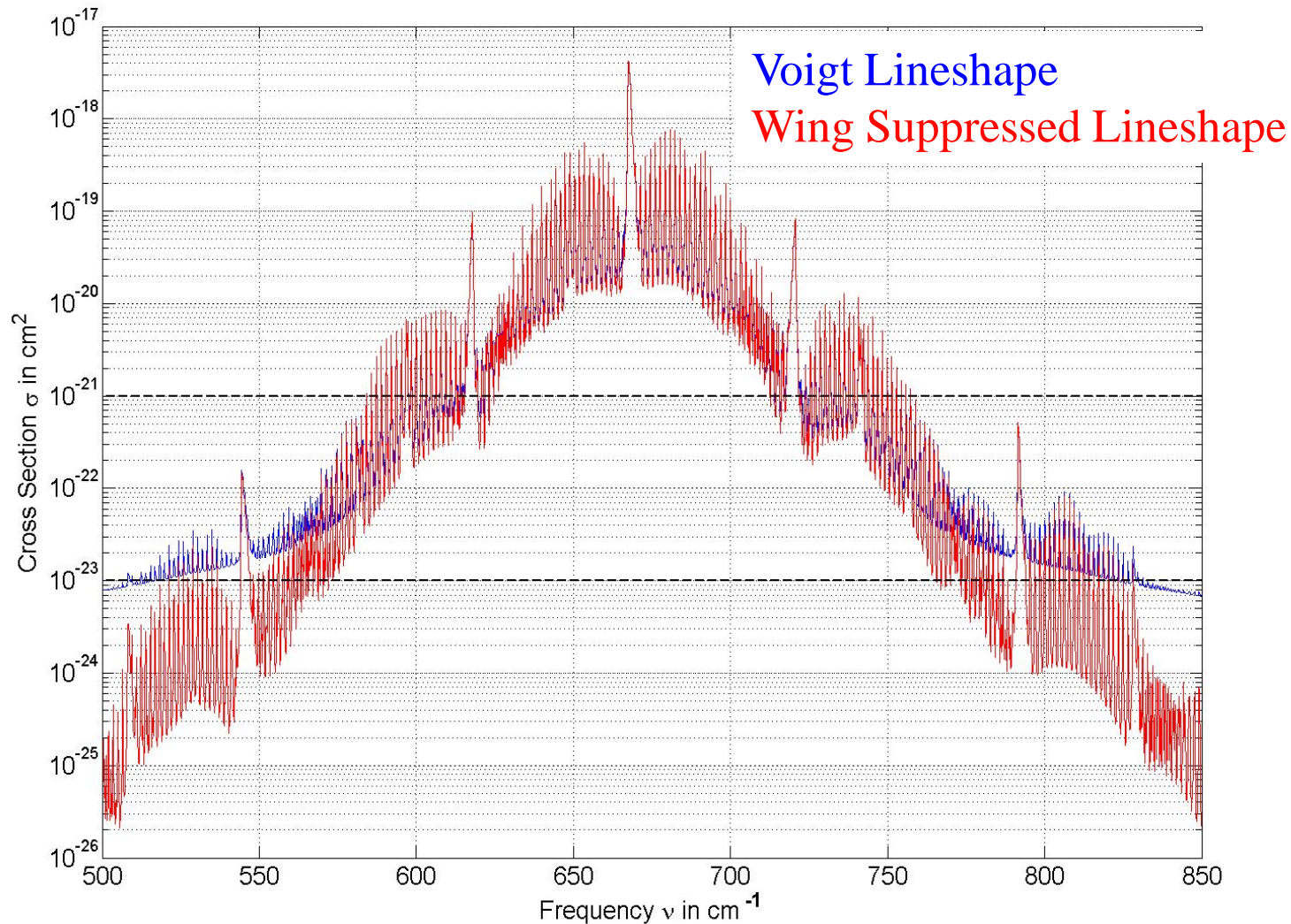


# CO<sub>2</sub> Bending Mode

at T = 300 K & P = 1 atm.

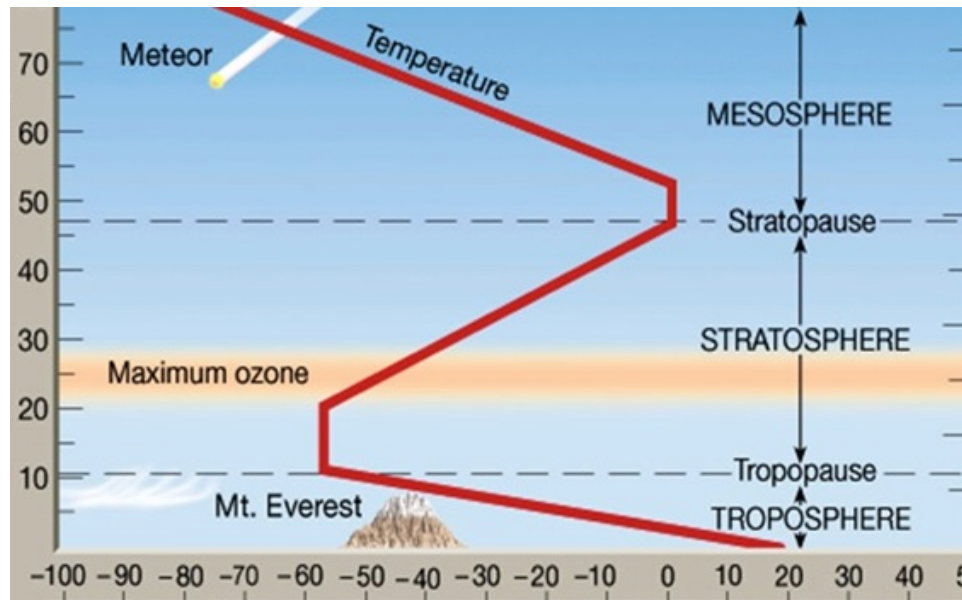


- Atmosphere opaque at line center → global warming driven by far wings of lineshape
- Far wings fall off faster than Voigt profile due to Dicke narrowing & collisions

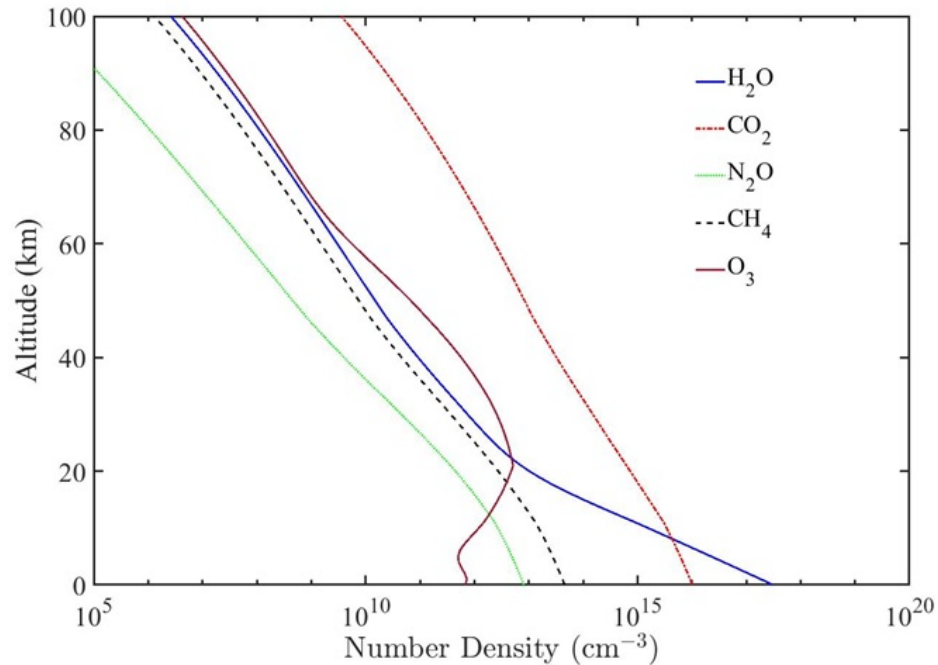


# Altitudinal Dependence

## Temperature



## Number Density



# Schwarzschild Equation

S. Chandrasekhar, *Radiative Transfer*, Dover, New York (1960)

Upward infrared heat intensity  $I(\nu, z)$  at frequency  $\nu$  & height  $z$  given by

$$\frac{\partial I(\nu, z)}{\partial z} = \kappa(B - I)$$

where attenuation  $\kappa = \sum_i N_i(z) \sigma_i(\nu, z)$



$\text{CO}_2, \text{N}_2\text{O}, \text{CH}_4, \text{H}_2\text{O}, \text{O}_3$

Planck Brightness  $B = \frac{2hc^2\nu^3}{e^{\nu/\nu_T} - 1} \quad \nu_T = \frac{kT}{hc}$

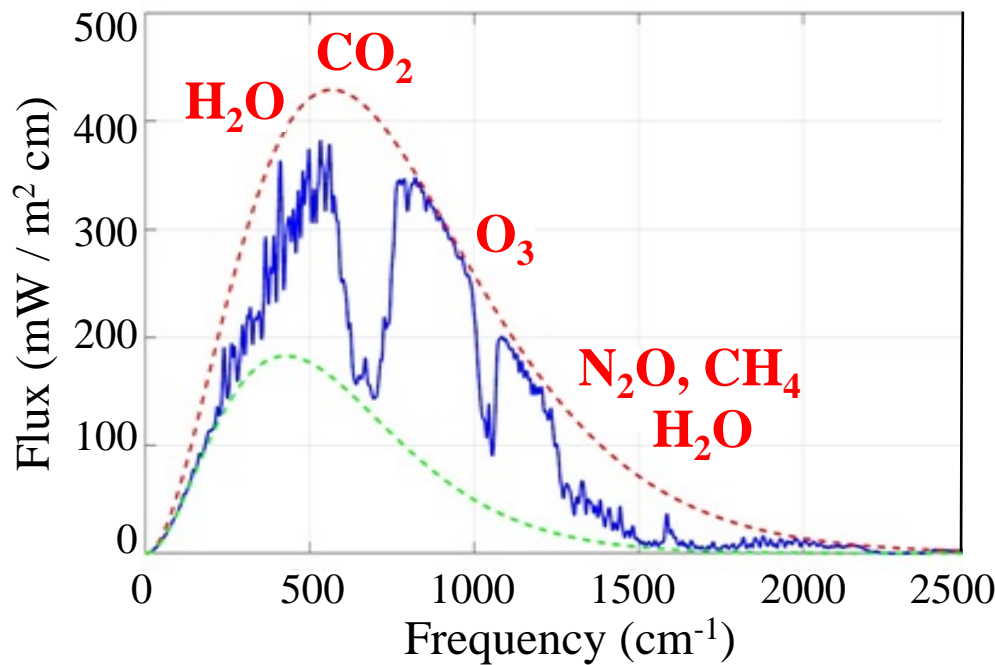
$$\pi \int_0^\infty B d\nu = \sigma T^4 \quad \text{Stefan Boltzmann Law}$$

# Comparison of Calculated Flux to Satellite Data

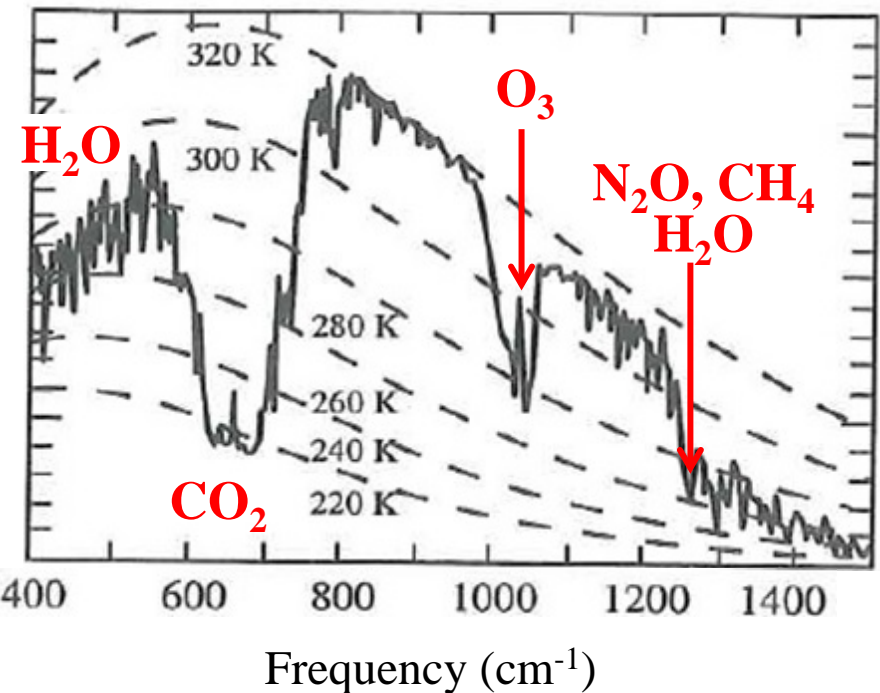
R. Hanel et al, *Geophys. Res.* 77, 2629 (1972)

- Flux calculation for ambient levels of  $\text{CO}_2$ ,  $\text{CH}_4$ ,  $\text{N}_2\text{O}$ ,  $\text{O}_3$  &  $\text{H}_2\text{O}$

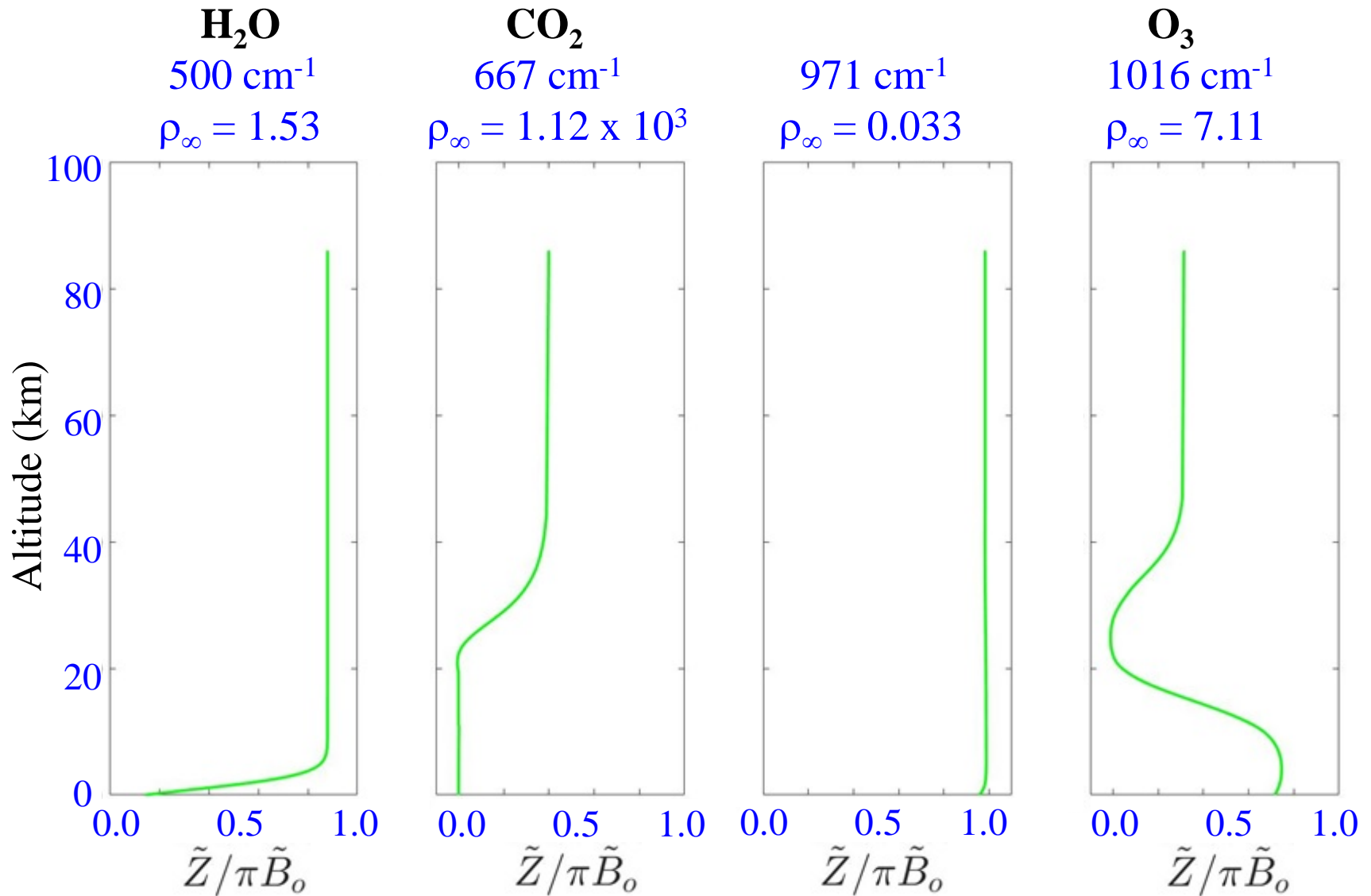
## Calculation



## Satellite Intensity Observations

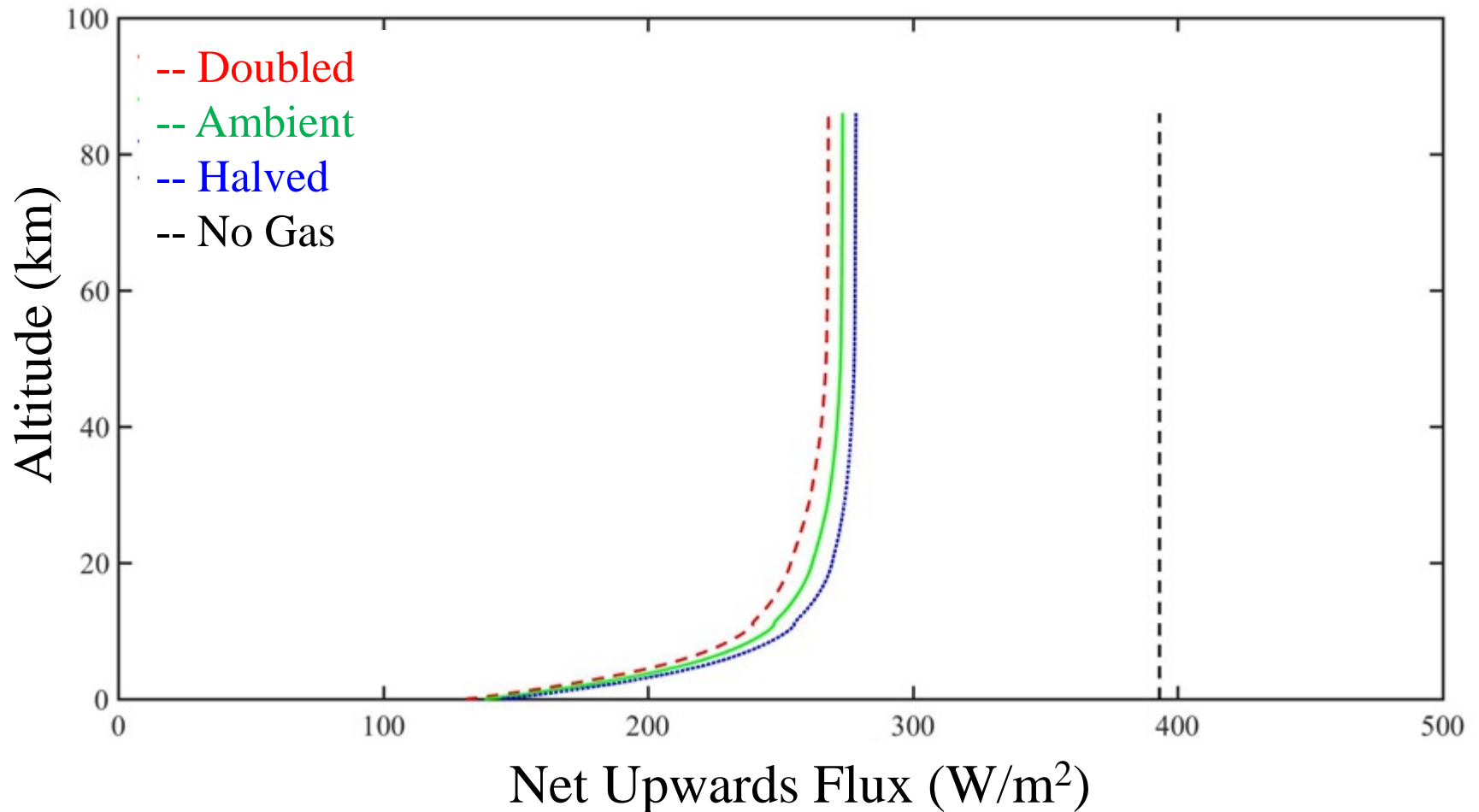


# Upwards Flux at various Frequencies



# Radiative Forcing

- Fluxes computed for surface at 288.7 K & standard atmosphere temperature profile
- Ambient Surface Levels: CO<sub>2</sub> (400 ppm), CH<sub>4</sub> (1.8 ppm ), N<sub>2</sub>O (0.32 ppm)
- Ambient Surface H<sub>2</sub>O (12,400 ppm due to 70% relative humidity) varied  $\pm 6\%$
- O<sub>3</sub> (7.8 ppm at 35 km) kept constant





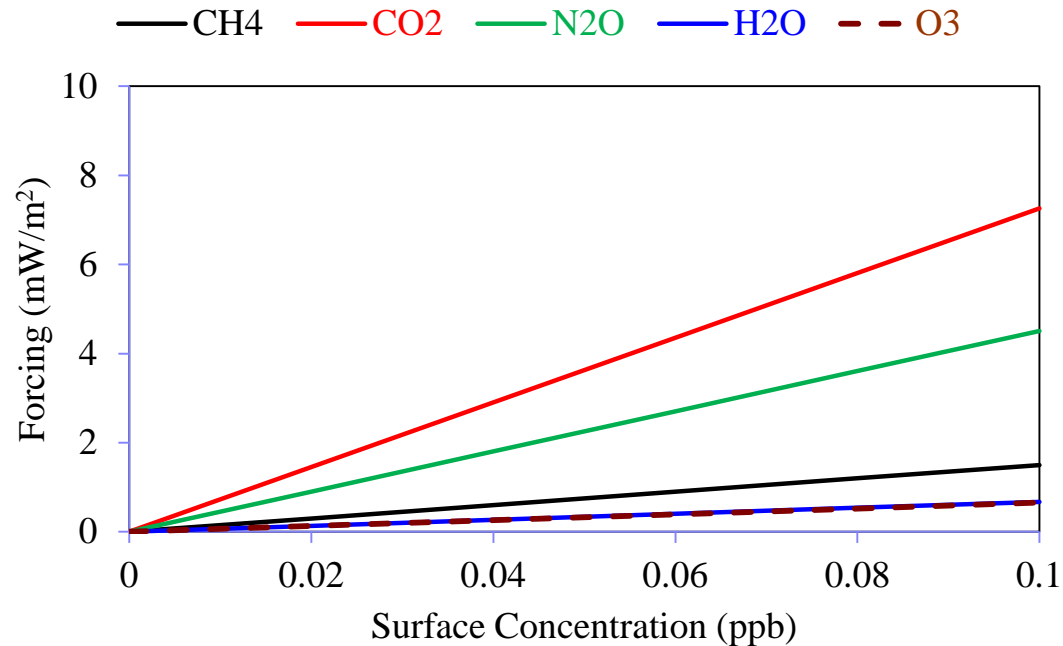
# Temperature Change due to Forcing

**Forcing:**  $\Delta J = J(\text{Ambient}) - J(\text{Doubled})$

**Temperature Increase:**  $\frac{\Delta T}{T} = \frac{\Delta J}{4 J}$

Gas	Concentration Change	Top of Atmosphere Forcing (W/m <sup>2</sup> )
H <sub>2</sub> O	+6% at 70% relative humidity	1.0
CO <sub>2</sub>	400 → 800 ppm	4.0
CH <sub>4</sub>	1.8 → 3.6 ppm	1.5
N <sub>2</sub> O	0.32 → 0.64 ppm	2.2
O <sub>3</sub>	No Change	
<b>Total</b>	<b>All of above</b>	<b>5.3</b> <b>⇒ ΔT = +1.4 °C</b>
<b>Plus</b>	<b>Clouds, Aerosols</b>	<b>????</b>

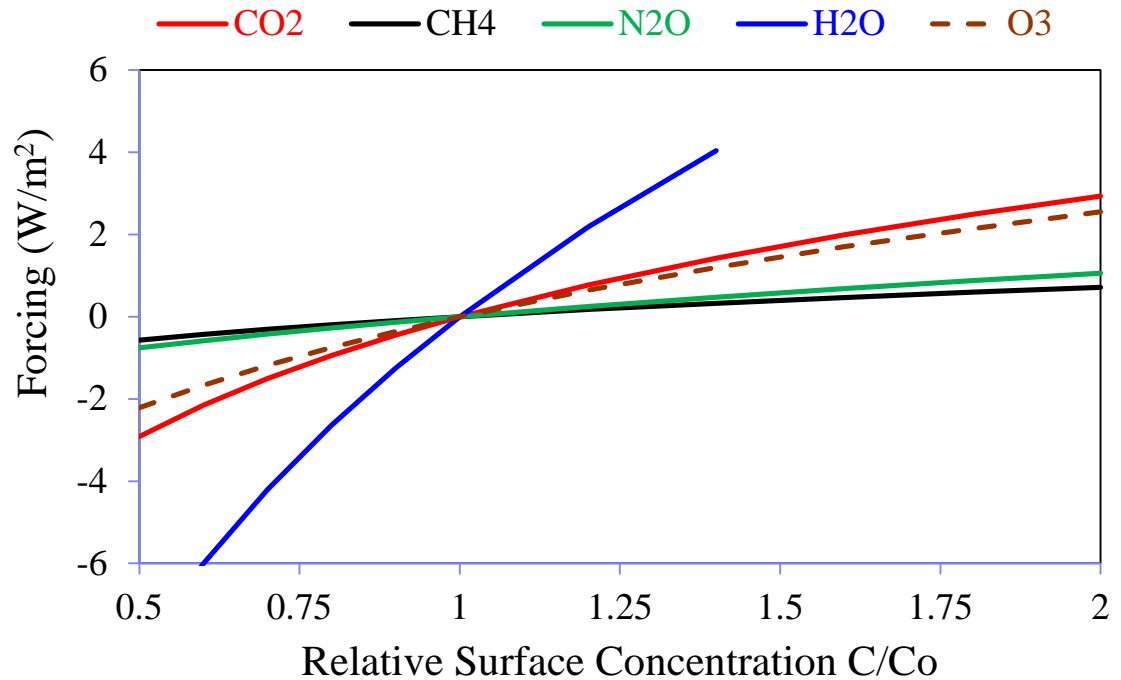
# Greenhouse Gas Potency for Barren Earth



Gas	0.1 ppb Forcing (mW/m <sup>2</sup> )	Column Density (10 <sup>19</sup> m <sup>-2</sup> )	Power/Molecule = Forcing/Col. Density (10 <sup>-22</sup> W)
CO <sub>2</sub>	7.3	2.16	3.5
N <sub>2</sub> O	4.5	2.07	2.2
CH <sub>4</sub>	1.5	2.09	0.7
H <sub>2</sub> O	0.7	0.50	1.4
O <sub>3</sub>	0.7	0.12	5.7

# Greenhouse Gas Potency for existing Atmosphere

Increase each greenhouse gas keeping other gas concentrations at ambient levels  $C_o$ .



Gas	0.1 ppb Forcing (mW/m <sup>2</sup> )	Column Density (10 <sup>19</sup> m <sup>-2</sup> )	Power/Molecule = Forcing/Col. Density (10 <sup>-22</sup> W)
CO <sub>2</sub>	1.1 x 10 <sup>-3</sup>	2.16	4.9 x 10 <sup>-4</sup>
N <sub>2</sub> O	4.1 x 10 <sup>-1</sup>	2.07	2.0 x 10 <sup>-1</sup>
CH <sub>4</sub>	5.1 x 10 <sup>-2</sup>	2.09	2.4 x 10 <sup>-2</sup>
H <sub>2</sub> O	9.8 x 10 <sup>-5</sup>	0.50	2.0 x 10 <sup>-4</sup>
O <sub>3</sub>	4.4 x 10 <sup>-2</sup>	0.12	3.8 x 10 <sup>-1</sup>

# Greenhouse Gas Potency Summary

WvW & W. Happer, submitted for publication

Molecule	Optically Thin Limit		Ambient Atmosphere	
	Power per Molecule ( $10^{-22}$ W)	Power rel. CO <sub>2</sub>	Power per Molecule ( $10^{-22}$ W)	Power rel. CO <sub>2</sub>
CO <sub>2</sub>	3.45	1	$4.91 \times 10^{-4}$	1
N <sub>2</sub> O	2.24	0.65	$1.96 \times 10^{-1}$	400
CH <sub>4</sub>	0.71	0.21	$2.44 \times 10^{-2}$	50
H <sub>2</sub> O	1.38	0.40	$1.96 \times 10^{-4}$	0.40
O <sub>3</sub>	5.67	1.64	$3.77 \times 10^{-1}$	770

# Conclusions

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## 1. Radiative Forcing

Doubling CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O & increasing H<sub>2</sub>O gives 5.3 W/m<sup>2</sup> forcing  
⇒ ~1.5 °C temperature increase. Uncertainty primarily due to clouds.

## 2. Greenhouse Potency

- At very low concentrations, CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O & H<sub>2</sub>O have comparable forcings
- At ambient levels CH<sub>4</sub> & N<sub>2</sub>O have greater per molecule forcing because CO<sub>2</sub> & H<sub>2</sub>O more strongly saturated due to 100-10,000 higher concentration

**More Info:** [www.wvanwijngaarden.info.yorku.ca](http://www.wvanwijngaarden.info.yorku.ca)